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Life This chapter covers the basics that you may have learned in your chemistry class. Whether your teacher goes over this chapter, or assigns it for you to review on your own, the questions that follow should help you focus on the most important points.

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protons; neutrons 2.
electrons 3. neutrons
4. electrons 5. ionic 6.
The two main types of
chemical bonds are
ionic and covalent
bonds.

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1. Introducing Biology.
Chapter 2: The
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Biology: McDougal
Littel. pages 34-64.
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18.3 Structure and
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Properties of the
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Occurrence

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composition of
isotopes, ion formation,
chemical bonds in ionic
and covalent
compounds, the types
of chemical reactions,
and the naming of
compounds.

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Chapter 2: Introduction
to the Chemistry of Life
Figure 2.1 Foods such
as bread, fruit, and
cheese are rich sources
of biological
macromolecules. The
elements carbon,
hydrogen, nitrogen,
oxygen, sulfur, and
phosphorus are the key
building blocks of the
chemicals found in

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living things.
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**Chapter 2: The
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(Part 1.3)**

Biology Chapter 2- The
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Essential Question:
What are the basic
chemical principles
that affect living
things?

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